



IUPAC Nomenclature of Coordination Complexes

(01) Write the IUPAC names of the following compounds.

- i. Mg_3N_2 -
- ii. FeCl_3 -
- iii. $\text{Na}_2\text{Cr}_2\text{O}_7$ -
- iv. $(\text{NH}_4)_2\text{Cr}_2\text{O}_7$ -
- v. $\text{Ni}(\text{MnO}_4)_2$ -

(02) Write the IUPAC names of the following ions.

- i. $[\text{CuCl}_4]^{2-}$ -
- ii. $[\text{CoCl}_4]^{2-}$ -
- iii. $[\text{Fe}(\text{CN})_6]^{4-}$ -
- iv. $[\text{Co}(\text{CN})_6]^{3-}$ -
- v. $[\text{Fe}(\text{CN})_6]^{3-}$ -
- vi. $[\text{Ag}(\text{CN})_2]^-$ -
- vii. $[\text{Cr}(\text{Br})_6]^{3-}$ -
- viii. $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$ -
- ix. $[\text{Cr}(\text{H}_2\text{O})_6]^{3+}$ -
- x. $[\text{Cr}(\text{H}_2\text{O})_2(\text{NH}_3)_4]^{3+}$ -
- xi. $[\text{Co}(\text{OH})(\text{NH}_3)_4(\text{H}_2\text{O})]^{2+}$ -
- xii. $[\text{Co}(\text{CN})_2(\text{NH}_3)_4]^+$ -
- xiii. $[\text{Fe}(\text{CN})_2(\text{NH}_3)_4]$ -
- xiv. $[\text{Fe}(\text{CN})_5\text{NO}]^{3-}$ -
- xv. $[\text{CoCl}(\text{NH}_3)_5]^{2+}$ -

(03) Write the IUPAC names of the following coordination complexes.

- i. K₄[Fe(CN)₆] -
- ii. K₃[Fe(CN)₅(NO)] -
- iii. Na₂[ZnCl₄] -
- iv. [Fe(SCN)₂]₂S -
- v. [CrCl₂(H₂O)₄]₂CrO₄ -
- vi. [Ag(NH₃)₂]Cl -
- vii. [Fe(OH)₂(H₂O)₄]Br -
- viii. [CoCl(NH₃)₅](NO₃)₂ -
- ix. [CoCl(NH₃)₅](NO₂)₂ -
- x. K₂[CuCl₄] -
- xi. [Cr(NH₃)₆][Fe(CN)₆] -
- xii. [Co(SO₄)(NH₃)₄]NO₃-
- xiii. Al[Ag(CN)₂]₃ -
- xiv. [Al(CN)₂]₂[CuF₄] -
- xv. [Fe(SCN)][CuCl₄] -

(04) Write the chemical formulae of the following complexes.

- i. Ammonium hexacyanoferrate (III) -
- ii. Tetraamminecopper (II) bromide -
- iii. Potassium hexacyanoferrate (III) -
- iv. Hexaamminecobalt(IV) tetrachlorocuprate(II) -
- v. Iron (III) hexacyanoferrate (II) -
- vi. Tetraquadichlorochromium (III) dichromate -
- vii. Dithiocyanatoiron (III) manganate -
- viii. Hexaaquanickel (II) tetrachlorocobaltate (II) -
- ix. Diamminebromidodicarbonylhydridocobalt(III) chloride -
- x. Pentaamminehydroxocobalt(III) nitrate -